



MATTER

Intelli Kid:

MATTER IS THE STUFF AROUND US

Matter is defined as anything that has mass and takes up space. Everything around us is matter like air, water, land, table, chair etc.





STATES OF MATTER

Liquids **Solids** Gases A solid has a definite A liquid has no definite A gas has no definite shape. shape. It takes the shape shape. of the container it is in. It takes the shape of the The molecules are The molecules are farther The molecules are much packed tightly together. apart. far apart. Can't be compressed. Can't be compressed. Can be compressed.

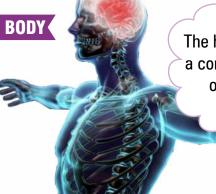






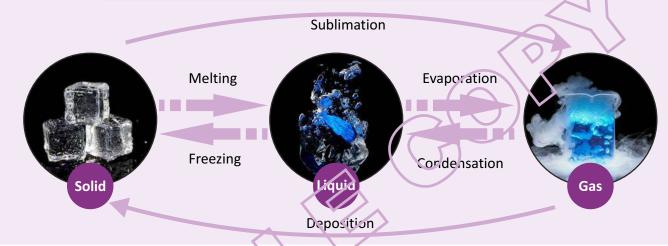
DIFFERENT STATES OF MATTER IN OUR BODY

In all of our bodies, you will find solid things like bones, teeth, skin, flesh etc., liquid, things like water, blood in our blood vessels and gaseous things like oxygen and carbon dioxide in our lungs.



The human body is a complex mixture of all states of matter.

INTERCONVERSION OF STATES OF MATTER



MELTING

Melting is the process by which a substance changes from the solid phase to the liquid phase.

EXAMPLE:

Melting of Ice-cream



FREEZING

Freezing is the process through which a substance changes from a liquid to a solid.

EXAMPLE:

Freezing of water to form ice.



EVAPORATION

Evaporation is the process of a substance in a liquid state changing to a gaseous state due to an increase in temperature.

EXAMPLE:

Evaporation of water from sea to form clouds.



CONDENSATION

Condensation is the change of the physical state of matter from the gas phase into the liquid phase, and is the reverse of evaporation.

EXAMPLE:

Water droplets forming on a glass having a cold drink.



SUBLIMATION

When anything solid turns into a gas without first becoming liquid, it is known as sublimation.

EXAMPLE:

Dry ice (Solid carbon dioxide) sublimates to form gas, similarly Naphthalene balls and Camphor are more examples.

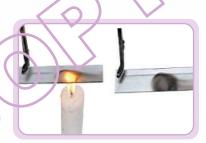


DEPOSITION

Changing a substance from gas state to solid state is called Deposition. It is reverse of Sublimation.

EXAMPLE:

Deposition of carbon due to candle flame.



SOLUBLE AND INSOLUBLE MATERIALS



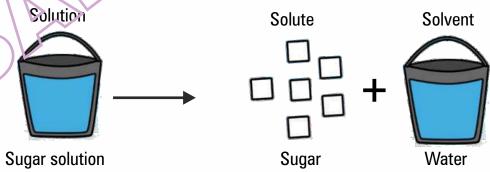
Solution:

A solution is basically two substances that are mixed together.

One of them is called the solute and the other is the solvent.

Solute is in less quantity & solvent is in large quantity.

Eg. - Salt solution, Sugar solution etc.



Soluble

The substance which dissolves in a liquid **Example**: Sugar, Salt, Medicine tablet etc.

Insoluble

The substance which does not dissolve in a liquid and forms precipitate Examples: Sand, Flour, Pebbles etc.





Q1. Fill in the table below by ticking the correct box. The first one has been done for you.

Objects	Keeps its shape	Keeps its volume	Can be compressed	Can't be compressed	Takes the shape of the container	Changes its volume	Occupies all the space of the container
		/		/	/		
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Q.2. Which of the following can form solutions? Tick (\checkmark) the correct ones.					
Salt, Cold drink	Salt, Wate	Salt, Water			
Pebbles, Water	☐ Thermocol	e balls, Water			
☐ Ink, Water	Lemon juic	Lemon juice, Cold Drink			
Sugar, Water	Lemon juic	☐ Lemon juice, Water			
Sand, Water	☐ Sugar, Mill	☐ Sugar, Milk			
Q.3. Colour Red for solute, Green for solvent and Blue for Solution.					
1. Coffee Powder	○ Milk	Coffee			
2. Rasna Powder	Rasna Drink	○ Water			
3. O Soda	Lemon juice	Fresh lime soda			
4. Rose syrup	Rose milk	Milk			
5. Tea power	Milk	○ Tea			
6. Surf Water	○ Water	Detergent			
7. Carbon dioxide	Fizzy Drinks	Water			
8. Mango Shake	Milk	Mango			





Q4. Tick (\checkmark) the correct option.

Everything in the universe is made up of $__1__$. Solids, liquids and gases are all made up of tiny particles called $__2__$. All forms of matter have $__3__$. The amount of space an object takes up is known as $__4__$. In liquids, this is often measured in $__5__$. Liquids take the $__6__$ of the container that holds them. The molecules of a solid are $__7__$. Gases mix together easily, a process that often creates $__8_$. The process of a liquid becoming a gas is known as $__9__$. When this happens, the molecules of the liquid $__10__$.					
1. Volume	☐ Matter	2.	Grams	Cells	
☐ Gases	☐ Fluid		Molecules	Electrons	
3. A smell	☐ Color	4.	Volume	☐ Weight	
A certain shape	☐ Mass		State	Property	
5. Meters Grams	Liters Molecules	6.	Weight Shape	☐ Mass	
7. Tightly packed Flying off into space	e [Loos Flow	ely packed ing		
8. Colors Vaporization	[Sme			
9. Condensation Fluidity	[Melt Vapo	ing orization		
10. Pack together Lose energy	[_	ff into space e close together		

EXERCISE

Intelli Quadros: Class-4
Matter



Q5. Match each term on the left with the best example on the right.

Volume

steam hits cold metal and forms water droplets

2 Solidifying

steam hits cold metal and forms water droplets

3 Evaporation

the temperature of liquid juice is lowered unt lit becomes a solid

4 Condensation

liquid molecules are moved to another container, where they take on a new shape

5 Boiling

cheese is heated on a stove until it becomes a liquid

6 Freezing

water from the ocean is heated by the sun and becomes water vapor

7 Pouring

liquid soup is heated in a microwave until it releases bubbles of water vapor



Q.6. Fill the blocks according to the directions given below. One has been done for you.

STATE	ADD	PROCESS	NEW STATE	EXAMPLE
	+ HEAT =	Sublimation	Gas	Sublimation of naphthalene balls
	+ = HEAT			
	COLD COLD			
	+			
	+ = HEAT			